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Cyclohexanespiro-2'-[2',3',6',7'-tetrahydro-1'H-cyclopenta[d]pyrimidin]-4'(5'H)-one

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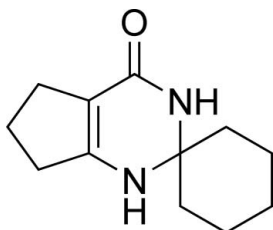
Received 7 January 2009; accepted 15 February 2009

Key indicators: single-crystal X-ray study; $T = 113$ K; mean $\sigma(\text{C}-\text{C}) = 0.002$ Å; R factor = 0.037; wR factor = 0.103; data-to-parameter ratio = 12.9.

The title compound, $\text{C}_{12}\text{H}_{18}\text{N}_2\text{O}$, was synthesized by the reaction of cyclohexanone and 2-aminocyclopent-1-enecarbonitrile. In the molecule of the title compound, the six-carbon ring displays a chair conformation, the six-membered 1,3-diazia ring and the cyclopentene ring both assume envelope conformations. Supramolecular aggregation is achieved by $\text{N}-\text{H}\cdots\text{O}$ hydrogen bonds.

Related literature

For general background on the biological activity of pyrimidinones, see: Schramm *et al.* (1984); Wen *et al.* (2002); For related structures, see: Yu *et al.* (1992); Zhang, Li, Shi *et al.* (2008); Zhang, Li, Yang *et al.* (2008).



Experimental

Crystal data

 $\text{C}_{12}\text{H}_{18}\text{N}_2\text{O}$ $M_r = 206.28$

Monoclinic, $P2_1/n$
 $a = 10.294$ (2) Å
 $b = 10.461$ (2) Å
 $c = 10.659$ (2) Å
 $\beta = 112.70$ (3)°
 $V = 1059.0$ (4) Å³

$Z = 4$
 Mo $K\alpha$ radiation
 $\mu = 0.08$ mm⁻¹
 $T = 113$ K
 $0.24 \times 0.20 \times 0.08$ mm

Data collection

Rigaku Saturn diffractometer
 Absorption correction: multi-scan
 (*CrystalClear*; Rigaku/MS, 2006)
 $T_{\min} = 0.980$, $T_{\max} = 0.993$

6976 measured reflections
 1862 independent reflections
 1632 reflections with $I > 2\sigma(I)$
 $R_{\text{int}} = 0.034$

Refinement

$R[F^2 > 2\sigma(F^2)] = 0.037$
 $wR(F^2) = 0.103$
 $S = 1.13$
 1862 reflections
 144 parameters
 2 restraints

H atoms treated by a mixture of independent and constrained refinement
 $\Delta\rho_{\text{max}} = 0.22$ e Å⁻³
 $\Delta\rho_{\text{min}} = -0.22$ e Å⁻³

Table 1

Hydrogen-bond geometry (Å, °).

$D-H\cdots A$	$D-H$	$H\cdots A$	$D\cdots A$	$D-H\cdots A$
$\text{N1}-\text{H1}\cdots\text{O1}^{\text{i}}$	0.899 (9)	1.993 (9)	2.8832 (14)	169.9 (14)
$\text{N2}-\text{H2}\cdots\text{O1}^{\text{ii}}$	0.899 (9)	2.080 (11)	2.9458 (17)	161.3 (15)

Symmetry codes: (i) $-x, -y + 1, -z$; (ii) $x + \frac{1}{2}, -y + \frac{1}{2}, z + \frac{1}{2}$.

Data collection: *CrystalClear* (Rigaku/MS, 2006); cell refinement: *CrystalClear*; data reduction: *CrystalClear*; program(s) used to solve structure: *SHELXS97* (Sheldrick, 2008); program(s) used to refine structure: *SHELXL97* (Sheldrick, 2008); molecular graphics: *SHELXTL* (Sheldrick, 2008); software used to prepare material for publication: *SHELXTL*.

We thank the Beijing Institute of Technology for financial support.

Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: PK2145).

References

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supplementary materials

Acta Cryst. (2009). E65, o615 [doi:10.1107/S1600536809005388]

Cyclohexanespiro-2'-[2',3',6',7'-tetrahydro-1'*H*-cyclopenta[*d*]pyrimidin]-4'(5'*H*)-one

D. Shi, D. Qian, Q. Zhang and J. Li

Comment

Pyrimidin-4(5*H*)-ones are valuable synthetic intermediates, and represent a common structure found in various biologically active compounds Schramm *et al.*, 1984). Functionalization of the pyrimidin-4(5*H*)-one group offers an attractive method for the generation of derivatives which may constitute interesting medicinal and biological properties. For example, spiro[cyclohexane-1,2'(1'*H*)-quinazolin]-4'(3'*H*)-one shows immunosuppressive, antifungal, and antitumor activity (Wen *et al.*, 2002).

Molecules of the title compound (Fig. 1) are linked by N1—H···O1 and N2—H···O1 H-bonds, as shown in Fig. 2 and have a similar conformation as (*s*)-2-(3-nitrophenyl)-1,2-dihydro-quinazolin-4(3*H*)-one (Zhang, Li & Shi *et al.*, 2008). The 1,3-diaza ring assumes envelope conformation, similar to that found in 4(3*H*)-quinazolinone derivatives (Zhang, Li & Yang *et al.*, 2008; Yu *et al.*, 1992). The cyclopentene exists in an envelope formation, and cyclohexane displays a chair conformation.

Experimental

A solution of 2-aminocyclohex-1-enecarbonitrile (10 mmol, 1.08 g) and sodium methanolate (10 mmol, 0.54 g) in cyclohexanone (2 ml), was refluxed for 2 h. The reaction mixture was cooled to 293 K and kept at this temperature for an additional 12 h. The solvent was filtered *in vacuo* to give 2-cyclohexyl-2,3,6,7-tetrahydro-1*H*-cyclopenta[*d*]pyrimidin-4(5*H*)-one. The product was recrystallized from ethanol to give clearless crystals. *M.p.* 513–514 K; IR (KBr): 3201 (N—H), 3074, 2931 (C—H), 1707 (C=O) cm⁻¹; ¹H-NMR(DMSO, p.p.m.): 1.50 (6*H*, m), 1.74–1.81 (4*H*, m), 2.28 (2*H*, t), 2.38 (2*H*, t), 2.50 (2*H*, m), 6.57 (1*H*,s), 6.86 (1*H*, s). 50 mg of the product was dissolved in ethyl acetate (5 ml) and the solution was kept at room temperature for 4 days to give colorless single crystals.

Refinement

H atoms attached to C were included in calculated positions with a riding model (C—H distance = 0.97 Å), while the N—H hydrogens were refined with N—H distance restraints of 0.90 Å. *U*_{iso} values were set to 1.2*U*_{eq} of the carrier atom.

Figures

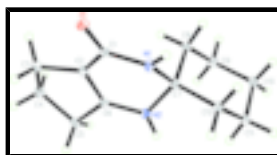


Fig. 1. The molecular structure of the title compound, drawn with 30% probability ellipsoids.

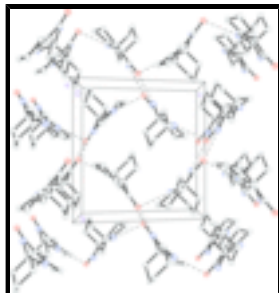


Fig. 2. The crystal structure of the title compound, viewed along *a* axis.

Cyclohexanespiro-2'-[2',3',6',7'-tetrahydro-1'*H*-cyclopenta[*d*]pyrimidin]-4'(5'*H*)-one

Crystal data

$C_{12}H_{18}N_2O$

$M_r = 206.28$

Monoclinic, $P2_1/n$

Hall symbol: -P 2yn

$a = 10.294$ (2) Å

$b = 10.461$ (2) Å

$c = 10.659$ (2) Å

$\beta = 112.70$ (3)°

$V = 1059.0$ (4) Å³

$Z = 4$

$F_{000} = 448$

$D_x = 1.294$ Mg m⁻³

Mo $K\alpha$ radiation

$\lambda = 0.71073$ Å

Cell parameters from 3035 reflections

$\theta = 2.8$ – 27.8 °

$\mu = 0.08$ mm⁻¹

$T = 113$ K

Plate, colorless

$0.24 \times 0.20 \times 0.08$ mm

Data collection

Rigaku Saturn
diffractometer

Radiation source: rotating anode

Monochromator: confocal multilayer optics

Detector resolution: 14.63 pixels mm⁻¹

$T = 113$ K

ω scans

Absorption correction: multi-scan
(CrystalClear; Rigaku/MSC, 2006)

$T_{\min} = 0.980$, $T_{\max} = 0.993$

6976 measured reflections

1862 independent reflections

1632 reflections with $I > 2\sigma(I)$

$R_{\text{int}} = 0.034$

$\theta_{\text{max}} = 25.0$ °

$\theta_{\text{min}} = 2.8$ °

$h = -12 \rightarrow 12$

$k = -12 \rightarrow 12$

$l = -12 \rightarrow 12$

Refinement

Refinement on F^2

Least-squares matrix: full

$R[F^2 > 2\sigma(F^2)] = 0.037$

$wR(F^2) = 0.103$

Secondary atom site location: difference Fourier map

Hydrogen site location: inferred from neighbouring sites

H atoms treated by a mixture of independent and constrained refinement

$w = 1/[\sigma^2(F_o^2) + (0.058P)^2 + 0.1813P]$

$S = 1.13$

1862 reflections

144 parameters

2 restraints

Primary atom site location: structure-invariant direct methods

where $P = (F_o^2 + 2F_c^2)/3$

$(\Delta/\sigma)_{\max} < 0.001$

$\Delta\rho_{\max} = 0.22 \text{ e } \text{Å}^{-3}$

$\Delta\rho_{\min} = -0.22 \text{ e } \text{Å}^{-3}$

Extinction correction: none

Special details

Geometry. All e.s.d.'s (except the e.s.d. in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell e.s.d.'s are taken into account individually in the estimation of e.s.d.'s in distances, angles and torsion angles; correlations between e.s.d.'s in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell e.s.d.'s is used for estimating e.s.d.'s involving l.s. planes.

Refinement. Refinement of F^2 against ALL reflections. The weighted R -factor wR and goodness of fit S are based on F^2 , conventional R -factors R are based on F , with F set to zero for negative F^2 . The threshold expression of $F^2 > \sigma(F^2)$ is used only for calculating R -factors(gt) *etc.* and is not relevant to the choice of reflections for refinement. R -factors based on F^2 are statistically about twice as large as those based on F , and R -factors based on ALL data will be even larger.

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters (Å^2)

	x	y	z	$U_{\text{iso}}^*/U_{\text{eq}}$
O1	-0.14346 (9)	0.42583 (8)	0.02403 (9)	0.0163 (2)
N1	0.08172 (11)	0.34945 (10)	0.09960 (11)	0.0152 (3)
N2	0.16234 (11)	0.18929 (10)	0.26816 (11)	0.0165 (3)
C1	-0.05332 (13)	0.34458 (11)	0.09213 (13)	0.0137 (3)
C2	-0.08039 (13)	0.24658 (11)	0.17362 (13)	0.0152 (3)
C3	-0.21300 (13)	0.22403 (12)	0.19860 (14)	0.0179 (3)
H3A	-0.2919	0.2044	0.1147	0.021*
H3B	-0.2364	0.2976	0.2412	0.021*
C4	-0.17249 (14)	0.10798 (13)	0.29541 (15)	0.0219 (3)
H4A	-0.2158	0.1143	0.3614	0.026*
H4B	-0.2031	0.0291	0.2447	0.026*
C5	-0.01109 (13)	0.11111 (13)	0.36688 (14)	0.0196 (3)
H5A	0.0185	0.1558	0.4531	0.023*
H5B	0.0282	0.0255	0.3820	0.023*
C6	0.03058 (13)	0.18298 (11)	0.26602 (13)	0.0152 (3)
C7	0.17877 (12)	0.24061 (11)	0.14698 (13)	0.0142 (3)
C8	0.14163 (13)	0.13822 (12)	0.03476 (14)	0.0165 (3)
H8A	0.0486	0.1045	0.0181	0.020*
H8B	0.1386	0.1776	-0.0487	0.020*
C9	0.24726 (14)	0.02825 (12)	0.07214 (14)	0.0206 (3)
H9A	0.2210	-0.0323	-0.0025	0.025*
H9B	0.2451	-0.0159	0.1513	0.025*
C10	0.39614 (13)	0.07713 (13)	0.10258 (15)	0.0212 (3)
H10A	0.4621	0.0064	0.1303	0.025*
H10B	0.4007	0.1147	0.0212	0.025*

supplementary materials

C11	0.43647 (13)	0.17692 (13)	0.21547 (14)	0.0194 (3)
H11A	0.5287	0.2112	0.2298	0.023*
H11B	0.4422	0.1364	0.2993	0.023*
C12	0.32970 (12)	0.28661 (12)	0.18128 (14)	0.0171 (3)
H12A	0.3561	0.3445	0.2582	0.021*
H12B	0.3333	0.3340	0.1045	0.021*
H1	0.1032 (15)	0.4130 (12)	0.0541 (14)	0.025 (4)*
H2	0.2347 (13)	0.1526 (15)	0.3359 (14)	0.034 (5)*

Atomic displacement parameters (\AA^2)

	U^{11}	U^{22}	U^{33}	U^{12}	U^{13}	U^{23}
O1	0.0151 (5)	0.0148 (4)	0.0169 (5)	0.0014 (4)	0.0038 (4)	0.0029 (4)
N1	0.0145 (6)	0.0128 (5)	0.0185 (6)	-0.0001 (4)	0.0066 (5)	0.0035 (5)
N2	0.0131 (6)	0.0219 (6)	0.0142 (6)	0.0028 (4)	0.0050 (5)	0.0048 (5)
C1	0.0142 (6)	0.0128 (6)	0.0121 (7)	-0.0016 (5)	0.0030 (5)	-0.0031 (5)
C2	0.0147 (6)	0.0144 (6)	0.0171 (7)	-0.0002 (5)	0.0067 (5)	0.0001 (5)
C3	0.0161 (6)	0.0178 (6)	0.0211 (7)	-0.0003 (5)	0.0087 (5)	0.0006 (6)
C4	0.0209 (7)	0.0242 (7)	0.0239 (8)	-0.0002 (6)	0.0125 (6)	0.0058 (6)
C5	0.0210 (7)	0.0214 (7)	0.0188 (8)	0.0034 (6)	0.0105 (6)	0.0054 (6)
C6	0.0178 (7)	0.0140 (6)	0.0152 (7)	-0.0009 (5)	0.0080 (5)	-0.0017 (5)
C7	0.0134 (6)	0.0146 (6)	0.0144 (7)	0.0003 (5)	0.0052 (5)	0.0021 (5)
C8	0.0161 (7)	0.0167 (6)	0.0157 (7)	-0.0015 (5)	0.0052 (5)	-0.0003 (5)
C9	0.0243 (7)	0.0160 (6)	0.0214 (8)	0.0009 (5)	0.0087 (6)	-0.0021 (5)
C10	0.0208 (7)	0.0234 (7)	0.0219 (8)	0.0052 (6)	0.0110 (6)	0.0003 (6)
C11	0.0134 (6)	0.0256 (7)	0.0197 (7)	0.0017 (5)	0.0068 (5)	-0.0002 (6)
C12	0.0151 (7)	0.0182 (6)	0.0190 (7)	-0.0021 (5)	0.0076 (5)	-0.0026 (5)

Geometric parameters (\AA , $^\circ$)

O1—C1	1.2612 (15)	C5—H5B	0.9700
N1—C1	1.3623 (16)	C7—C12	1.5299 (17)
N1—C7	1.4698 (16)	C7—C8	1.5399 (18)
N1—H1	0.899 (9)	C8—C9	1.5265 (18)
N2—C6	1.3494 (17)	C8—H8A	0.9700
N2—C7	1.4676 (17)	C8—H8B	0.9700
N2—H2	0.899 (9)	C9—C10	1.5271 (19)
C1—C2	1.4387 (17)	C9—H9A	0.9700
C2—C6	1.3599 (18)	C9—H9B	0.9700
C2—C3	1.5068 (17)	C10—C11	1.5246 (19)
C3—C4	1.5429 (19)	C10—H10A	0.9700
C3—H3A	0.9700	C10—H10B	0.9700
C3—H3B	0.9700	C11—C12	1.5324 (18)
C4—C5	1.5380 (19)	C11—H11A	0.9700
C4—H4A	0.9700	C11—H11B	0.9700
C4—H4B	0.9700	C12—H12A	0.9700
C5—C6	1.5036 (18)	C12—H12B	0.9700
C5—H5A	0.9700		

C1—N1—C7	122.29 (10)	N1—C7—C12	109.36 (10)
C1—N1—H1	117.0 (9)	N2—C7—C8	110.53 (10)
C7—N1—H1	118.8 (9)	N1—C7—C8	109.86 (10)
C6—N2—C7	117.35 (11)	C12—C7—C8	109.22 (10)
C6—N2—H2	120.6 (11)	C9—C8—C7	112.53 (11)
C7—N2—H2	121.4 (11)	C9—C8—H8A	109.1
O1—C1—N1	121.07 (11)	C7—C8—H8A	109.1
O1—C1—C2	123.83 (11)	C9—C8—H8B	109.1
N1—C1—C2	114.98 (11)	C7—C8—H8B	109.1
C6—C2—C1	118.79 (11)	H8A—C8—H8B	107.8
C6—C2—C3	111.14 (11)	C8—C9—C10	111.01 (11)
C1—C2—C3	128.16 (11)	C8—C9—H9A	109.4
C2—C3—C4	102.23 (10)	C10—C9—H9A	109.4
C2—C3—H3A	111.3	C8—C9—H9B	109.4
C4—C3—H3A	111.3	C10—C9—H9B	109.4
C2—C3—H3B	111.3	H9A—C9—H9B	108.0
C4—C3—H3B	111.3	C11—C10—C9	110.02 (11)
H3A—C3—H3B	109.2	C11—C10—H10A	109.7
C5—C4—C3	106.03 (10)	C9—C10—H10A	109.7
C5—C4—H4A	110.5	C11—C10—H10B	109.7
C3—C4—H4A	110.5	C9—C10—H10B	109.7
C5—C4—H4B	110.5	H10A—C10—H10B	108.2
C3—C4—H4B	110.5	C10—C11—C12	111.91 (11)
H4A—C4—H4B	108.7	C10—C11—H11A	109.2
C6—C5—C4	102.01 (11)	C12—C11—H11A	109.2
C6—C5—H5A	111.4	C10—C11—H11B	109.2
C4—C5—H5A	111.4	C12—C11—H11B	109.2
C6—C5—H5B	111.4	H11A—C11—H11B	107.9
C4—C5—H5B	111.4	C7—C12—C11	112.98 (10)
H5A—C5—H5B	109.2	C7—C12—H12A	109.0
N2—C6—C2	123.11 (12)	C11—C12—H12A	109.0
N2—C6—C5	125.07 (11)	C7—C12—H12B	109.0
C2—C6—C5	111.77 (11)	C11—C12—H12B	109.0
N2—C7—N1	106.97 (10)	H12A—C12—H12B	107.8
N2—C7—C12	110.86 (11)		
C7—N1—C1—O1	-163.93 (11)	C4—C5—C6—C2	16.59 (14)
C7—N1—C1—C2	19.87 (17)	C6—N2—C7—N1	40.32 (14)
O1—C1—C2—C6	-165.34 (12)	C6—N2—C7—C12	159.48 (11)
N1—C1—C2—C6	10.73 (17)	C6—N2—C7—C8	-79.26 (13)
O1—C1—C2—C3	-2.6 (2)	C1—N1—C7—N2	-44.20 (15)
N1—C1—C2—C3	173.52 (12)	C1—N1—C7—C12	-164.32 (11)
C6—C2—C3—C4	-15.01 (14)	C1—N1—C7—C8	75.81 (14)
C1—C2—C3—C4	-178.87 (13)	N2—C7—C8—C9	-67.76 (13)
C2—C3—C4—C5	24.67 (14)	N1—C7—C8—C9	174.42 (10)
C3—C4—C5—C6	-25.13 (14)	C12—C7—C8—C9	54.46 (14)
C7—N2—C6—C2	-15.31 (18)	C7—C8—C9—C10	-57.49 (15)
C7—N2—C6—C5	167.51 (12)	C8—C9—C10—C11	56.57 (15)
C1—C2—C6—N2	-12.96 (19)	C9—C10—C11—C12	-55.28 (15)
C3—C2—C6—N2	-178.51 (11)	N2—C7—C12—C11	69.13 (14)

supplementary materials

C1—C2—C6—C5	164.56 (11)	N1—C7—C12—C11	-173.17 (11)
C3—C2—C6—C5	-1.00 (15)	C8—C7—C12—C11	-52.91 (15)
C4—C5—C6—N2	-165.96 (12)	C10—C11—C12—C7	54.85 (15)

Hydrogen-bond geometry (Å, °)

<i>D</i> —H \cdots <i>A</i>	<i>D</i> —H	H \cdots <i>A</i>	<i>D</i> \cdots <i>A</i>	<i>D</i> —H \cdots <i>A</i>
N1—H1 \cdots O1 ⁱ	0.899 (9)	1.993 (9)	2.8832 (14)	169.9 (14)
N2—H2 \cdots O1 ⁱⁱ	0.899 (9)	2.080 (11)	2.9458 (17)	161.3 (15)

Symmetry codes: (i) $-x, -y+1, -z$; (ii) $x+1/2, -y+1/2, z+1/2$.

Fig. 1

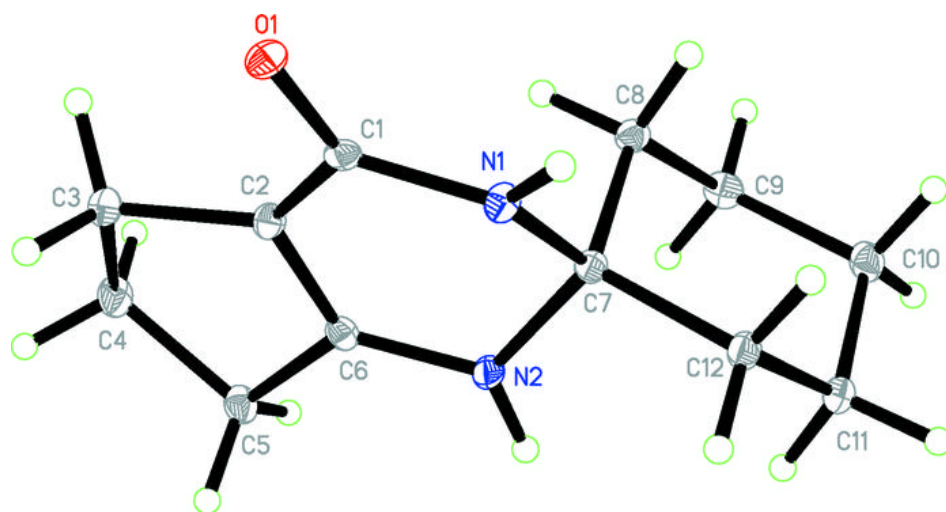


Fig. 2

